

Asymmetric Amination of Tetralone and Chromanone Derivatives Employing ω -Transaminases

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Supporting Information

ABSTRACT: Various (S)-selective and (R)-selective ω -transaminases were investigated for the amination of 1- and 2-tetralone R and derivatives as well as of 3- and 4-chromanone. All ketones tested were aminated to give the corresponding enantiopure amines (*ee* > 99%) employing at least one of the enzymes investigated. In most of the cases the (S)- as well as the (R)-



enantiomer was obtained in optically pure form. The amination of 3-chromanone was performed on a 100 mg scale leading to optically pure (R)-3-aminochromane (ee > 99%) with complete conversion and 78% isolated yield.

KEYWORDS: asymmetric catalysis, biocatalysis, amination, ω -transaminase, aminochromane, aminotetraline

INTRODUCTION

Chiral amine moieties are ubiquitously found in a vast number of natural products and bioactive compounds, like pharmaceuticals and agro chemicals.^{1–5} For instance, the 1-aminotetraline backbone can be found in various pharmaceuticals such as the antidepressiva Sertraline^{6–11} or Norsertraline.¹² 2-Aminotetraline represents the scaffold of Rotigotine, a drug employed for the treatment of Parkinson as well as Willis-Ekbom diseases.¹³ Furthermore, (S)-3-aminochromane shares the same backbone as the antidepressant and anxiolytic drug Alnespirone,¹⁴ whereas its enantiomeric form is a part of the antihypertensive Etamicastat^{15,16} and the 5-HT1A receptor antagonist Robalzotan.^{17,18}

Aminotetraline and aminochromane derivatives have been obtained previously by resolution of racemic mixtures using fractional crystallization^{19–22} or chiral auxiliaries¹² as well as via organic synthesis from chiral precursors.^{23–25} Dynamic kinetic resolution (DKR) (i.e., Ru, Pd, Ni catalyzed racemization combined with a lipase catalyzed resolution) afforded the amines in high optical purity (>99% *ee*) and good yields.^{26,27} The DKR of amines was plagued by the low racemization rate (up to 3 days) even at higher temperature (70 °C) and the often not perfect chemoselectivity.^{28–30} Adding gaseous hydrogen (ca. 0.1 bar)³¹ or applying microwave irradiation accelerated the racemization rate.³² The asymmetric hydrogenation using metal-catalysts at high pressure (i.e., up to 100 bar) of enamides, imines, and oximes furnished optically active amines.^{1–3,33} Interestingly, the asymmetric hydrogenation of the enamide analogues of the ketones **2b** and **2e** led to high *ee*'s (98% and >99%, respectively),^{34–37} whereas the enamides of **2a** and **2f** were hydrogenated with 92–94% *ee*.^{38–40}

A metal-free biocatalytic amination of prochiral ketones can be performed employing ω -transaminases (ω TAs).^{1,41-45} ω TAs have been recently investigated with increased efforts for asymmetric synthesis,⁴⁶⁻⁶³ leading also to the launch of an industrial process.⁶⁴

RESULTS AND DISCUSSION

Various (S)-selective and (R)-selective ω -transaminases were investigated for the asymmetric amination of 1- and 2-tetralone and derivatives as well as of 3- and 4-chromanone. The ω TAs were employed as freeze-dried cell powder of *E. coli* containing the overexpressed enzyme. The amination of the ketones was conducted using alanine as amine donor and removing the coproduct pyruvate by either a lactate dehydrogenase (LDH, Scheme 1, LDH-system) or alanine dehydrogenase (AlaDH, AlaDH-system). The cofactor NADH required by the LDH/ AlaDH was recycled employing GDH-glucose or FDH-formate. Depending on the stereoselectivity of the ω TA, either L- or Dalanine was supplied.

As an alternative, the bioamination was performed using 2propylamine as amine donor employing the engineered variant from *Arthrobacter* sp. $(ArRmut11-\omega TA)^{64}$ in aqueous phase at pH 11 leading to acetone as a coproduct (Scheme 2, 2-PrNH₂ system). This variant was designed for the amination of sterically hindered ketones bearing to bulky substituents (bulky-bulky ketones) on an industrial scale.⁶⁴ Dimethylsulfoxide (DMSO) was added as cosolvent in few selected cases (20% v v⁻¹) for transformations employing (*R*)-selective ω TAs.

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Scheme 1. Asymmetric Reductive Amination of Prochiral Ketones (1a-f) Employing ω -Transaminases (ωTAs) with Alanine as Amine Donor



Scheme 2. Asymmetric Bioamination Employing 2-Propylamine as Amine Donor (2-PrNH₂ System)



The (S)-selective ω TAs from the following strains were tested: Vibrio fluvialis (VF- ω TA),⁶⁵ Chromobacterium violaceum (CV- ω TA),⁶⁶ Bacillus megaterium (BM- ω TA),⁶⁷ Paracoccus denitrificans (PD- ω TA),⁶⁸ Pseudomonas fluorescens (PF- ω TA)⁶⁹ as well as a variant from Arthrobacter citreus (ArS- ω TA).⁷⁰ The (R)-selective ω TAs originated from Aspergillus terreus (AT- ω TA),⁵⁹ Hyphomonas neptunium (HN- ω TA),⁵⁹ Arthrobacter sp. (ArR- ω TA)⁷¹ as well as its variant ArRmut11- ω TA.⁶⁴

 β -Tetralone 1a was in general very well accepted by most of the (S)-selective transaminases leading to high or even complete conversion. Interestingly, only ArS- ω TA performed the amination with ee > 99% (Table 1, entry 1). The (R)selective ArR- ω TAs and the variant from Arthrobacter sp. (ArRmut11- ω TA) aminated ketone 1a yielding the amine with only moderate ee's (76 and 54%, entries 3–4), while AT- ω TA and HN- ω TA were not active at all.

Introducing a substituent in 6-position, 6-methoxy substituted β -tetralone **1b** was aminated with slightly higher stereoselectivities compared to **1a** by various ω TAs (see Supporting Information). For instance, (S)-6-methoxy-2aminotetraline **2b** was obtained in enantiopure form using ArS- ω TA as well as BM- ω TA (entries 5–6) while ArRmut11- ω TA yielded (R)-**2b** with perfect stereoselectivity (>99% *ee*, entry 8). On the other hand CV- ω TA led to (S)-**2b** with moderate *ee* (79%, entry 7). The stereoselectivity of CV improved significantly when the methoxy substituent was in 7position (substrate **1c**). Actually five (S)-selective ω TAs out of

Table 1. Selected Results for the Asymmetric Amination of1- and 2-Tetralones and Derivatives

entry	ωTA	substr.	system ^a	conv. $(\%)^b$	ee (%) ^c
1	ArS	1a	LDH	>99	>99 (S)
2	CV	1a	LDH	>99	74 (S)
3	ArRmut11	1a	2-PrNH2	>99	54 (R)
4	ArR	1a	AlaDH	14	76 (R)
5	ArS	1b	AlaDH	90	>99 (S)
6	BM	1b	AlaDH	88	>99 (S)
7	CV	1b	LDH	67	79 (S)
8	ArRmut11	1b	2-PrNH2	>99	>99 (R)
9	VF	1c	AlaDH/LDH	>99	>99 (S)
10	PD	1c	AlaDH/LDH	>99	>99 (S)
11	CV	1c	AlaDH/LDH	>99	>99 (S)
12	PF	1c	AlaDH/LDH	>99	>99 (S)
13	ArRmut11	1c	2-PrNH2	>99	>99 (R)
14	VF	1d	LDH	22	98 (S)
15	PD	1d	LDH	8	>99 (S)
16	ArRmut11	1d	2-PrNH2	26	$>99 (R)^{56}$

^aSubstrate (50 mM), freeze-dried *E. coli* cells containing overexpressed ω TA (20 mg), LDH-system: L-/D-alanine (250 mM), phosphate buffer (pH 7, 100 mM), PLP (1 mM), LDH (90 U), GDH (30 U), NAD⁺ (1 mM), glucose (150 mM). AlaDH-system: L-/D-alanine (250 mM), phosphate buffer (pH 7, 100 mM), PLP (1 mM), AlaDH (12 U), FDH (11 U), NAD⁺ (1 mM), ammonium formate (150 mM). <u>2-PrNH₂ system</u>: water, 2-propylamine (1 M), PLP (0.5 mM), adjusted to pH 11 with aqueous 3 N HCl. ^b24 h reaction time, 30 °C, shaking 120 rpm. ^cDetermined by GC on a chiral stationary phase.

six, namely, VF, PD, BM, CV, and PF converted (S)-7methoxy-2-tetralone (1c) with perfect *ee* (>99%). Additionally quantitative conversion could be achieved in four cases (>99%, entry 10–13). ArRmut11- ω TA was also for this substrate the only applicable (*R*)-selective ω TA giving (*R*)-2c with perfect conversion and stereoselectivity (>99% conv., >99% *ee*, entry 13).

Although 1-tetralone 1d led in general to low conversions, the analogous 4-chromanone 1e was transformed with higher conversion using (S)-selective as well as (R)-selective ω TAs (Table 2). For instance, VF-, PD-, and BM- ω TA aminated 1e with perfect stereoselectivity (*ee* > 99%) (entries 2–4). The mirror image (R)-2e was also obtained with perfect *ee* (>99%) and 54% conversion using ArRmut11- ω TA (entry 5).

The amination of 3-chromanone 1f led to the formation of (R)-3-aminochromane (R)-2f employing the (S)-selective ω TAs due to a switch in the CIP priority. Particularly, ArS- ω TA afforded (R)-2f in optically pure form (>99% ee) and quantitative conversion (entry 7). (R)-2f is described as a building block for a medication for treating chronic heart disorders.⁷¹ All other (S)-selective ω TAs showed lower stereoselectivity (see Supporting Information). Surprisingly, a switch of the stereopreference for the amination of 1f was observed for ArRmut11- ω TA as well as HN- ω TA. Both enzymes led to the formation of (R)-2f, thus the same enantiomer as obtained with the (S)-selective ω TAs: HN- ω TA even gave (R)-2f in optically pure form (>99% ee, entry 8), whereas ArRmut11- ω TA led to 67% ee (R). In contrast, AT- ω TA retained the expected stereopreference yielding (S)-2f with 56% ee and excellent conversion (94%) when DMSO was added as cosolvent (20% v v^{-1}) (entry 10).

Finally, the bioamination of 1f with ArS- ω TA was shown on a 100 mg scale leading to (*R*)-2f with full conversion and perfect optical purity (*ee* > 99%); after extraction the amine

 Table 2. Selected Results for the Asymmetric Amination of

 3- and 4-Chromanone

entry	ωTA	substr.	system ^a	conv. $(\%)^b$	ee (%) ^c
1	PF	1e	AlaDH	13	85 (S)
2	VF	1e	LDH	56	>99 (S)
3	PD	1e	LDH	45	>99 (S)
4	BM	1e	AlaDH	11	>99 (S)
5	ArRmut11	1e	2-PrNH ₂	54	>99 (R)
6	ArS	1e	AlaDH	25	88 (S)
7	ArS	1f	LDH	>99	>99 $(R)^{d}$
8	HN	1f	AlaDH	22	>99 $(R)^{d,e}$
9	ArRmut11	1f	2-PrNH ₂	>99	67 $(R)^{d,e}$
10	AT	1f	LDH	94 ^f	56 $(S)^{d}$

^aSubstrate (50 mM), freeze-dried *E. coli* cells containing overexpressed ω TA (20 mg); LDH System: L-/D-alanine (250 mM), phosphate buffer (pH 7, 100 mM), PLP (1 mM), LDH (90 U), GDH (30 U), NAD⁺ (1 mM), glucose (150 mM). AlaDH-system: L-/D-alanine (250 mM), phosphate buffer (pH 7, 100 mM), PLP (1 mM), AlaDH (12 U), FDH (11 U), NAD⁺ (1 mM), ammonium formate (150 mM). <u>2</u>-<u>PrNH₂</u> system: reaction conditions: water, 2-propylamine (1 M), PLP (0.5 mM), adjusted to pH 11 with aqueous 3 N HCl. ^b24 h reaction time, 30 °C, shaking 120 rpm. ^cDetermined by GC on a chiral stationary phase. ^dSwitch in CIP priority compared to amines **2a**–e. ^eHN- and ArRmut11- ω TA displayed opposite stereopreference than expected. ^f20% v v⁻¹ DMSO employed as cosolvent.

product was precipitated and isolated as its hydrochloride salt (R)-2f-HCl with 78% isolated yield.

In summary, various (S)- and (R)-selective ω -transaminases were employed for the asymmetric reductive amination of 1and 2- tetralone and derivatives as well as 3- and 4chromanone. In all the cases investigated, the (S)-selective ω transaminase enabled amine formation in enantiopure form using at least one of the tested (S)- ω TAs. The in general (R)selective transaminases HN- ω TA and ArRmut11 displayed for 3-chromanone If an inverted stereopreference compared to the other substrates in this paper as well as to published substrates; thus it led for this single substrate to the same amine enantiomer as the (S)-selective transaminases. The corresponding (R)-enantiomers were also accessible in optically pure form in most of the cases, the only exception being substrates 1a. The bioamination was successfully exemplified on a preparative scale (100 mg) for substrate 1f.

EXPERIMENTAL SECTION

Ketones 1a-e, racemic amine rac-2d and enantiopure amine (R)-2d were commercially available from Sigma-Aldrich (Vienna, Austria). Racemic amines rac-2a,c,e and enantiopure amines (S)-2c, (S)-2d, and (R)-2d were a gift from BASF. Ketone 1f was synthesized according to literature.^{72,73} Racemic amine rac-2b was synthesized according to literature.^{72,73} Solvents were purchased from Roth and used as received unless otherwise stated. PureLinkTM Quick Plasmid Minipreparation Kit (K2100-10) and One Shot TOP10 chemically competent E. coli cells (SKU# C6000-03) were obtained from Invitrogen (Karlsruhe, Germany). L-Lactate dehydrogenase from rabbit muscle (lyophilized powder, 136 U mg⁻¹ protein, one unit will reduce 1.0 μ mol of pyruvate to L-lactate per min at pH 7 at 25 °C, catalog no. 61309) was purchased from Sigma-Aldrich (Vienna, Austria). Glucose dehydrogenase (lyophilized powder, 25 U mg⁻¹, one unit will oxidize 1 μ mol of β -D-glucose

to D-glucono- δ -lactone per min at pH 8.0 and 37 °C, catalog no. B-4) was purchased from X-zyme (Düsseldorf, Germany). Formate dehydrogenase from *Candida boidinii* (lyophilized crude enzyme, 2.2 U mg⁻¹, one unit will oxidize 1.0 μ mole of formate to CO₂ per min at pH 7.6 at 37 °C, catalog no. FDH-101) and β -NAD free acid were purchased from Codexis (Redwood City, CA, U.S.A.).

The conversions to amines were measured by GC using an Agilent 7890 A GC system, equipped with a FID detector and using an Agilent J&W DB-1701 column (30 m, 250 μ m, 0.25 μ m). Helium was used as carrier gas and EtOAc was used as solvent. The *ee*'s of amines were measured by GC using an Agilent 7890 A GC system, equipped with a FID detector and using a Varian Chrompack Chirasel Dex-CB column (25 m, 320 μ m, 0.25 μ m). Hydrogen was used as carrier gas. All GC-MS measurements were carried out with an Agilent 7890A GC system, equipped with an Agilent 5975C mass-selective detector (electron impact, 70 eV) and using an Agilent HP-5 ms capillary column (5%-phenyl-methylpolysiloxane phase, 30 m, 250 μ m, 0.25 μ m). Helium was used as carrier gas.

Optical rotations were measured with a Perkin-Elmer Polarimeter 341 in a 1 mL cuvette of 10 cm length. NMR spectra were recorded on a Bruker 360 MHz spectrometer using TMS as internal standard.

L-alanine was used for the amination employing the (S)-selective ω TAs, while D-alanine was employed for (R)-selective ω TAs.

Lyophilized *E. coli* cells containing overexpressed ω -transaminases were prepared as previously reported but with elongated induction time of the enzyme (16 h).^{51,56,74} Purified recombinant L-alanine dehydrogenase was prepared as described recently.⁵⁶

Representative Example for Reductive Amination Employing the AlaDH-System. Lyophilized cells of *E. coli* BL21 (DE3) containing overexpressed ω -transaminases (20 mg) were rehydrated in an Eppendorf tube (2 mL) in phosphate buffer (0.6 mL, 100 mM, pH 7, 1 mM PLP) for 30 min at 30 °C and 120 rpm on an orbital shaker turned 90° sideward. Phosphate buffer (0.4 mL, 100 mM, pH 7, 1 mM PLP) containing ammonium formate (9.5 mg, final concentration 150 mM), alanine (22.3 mg, final concentration 250 mM), formate dehydrogenase (11 U), and NAD⁺ (0.67 mg, final concentration 1 mM), as well as the ketone (50 mM) and L-alanine dehydrogenase (12 U) were added.

L- or D-alanine were used as amine donor employing either (S)- or (R)-selective ω TAs, respectively. DMSO was added in selected experiments using (R)-selective ω TAs (15% v v⁻¹). The mixture was shaken at 30 °C and 120 rpm. The reaction was quenched after 24 h by the addition of aqueous NaOH (200 μ L, 10 N), and the reaction mixture was extracted with ethyl acetate (2 × 500 μ L). The combined organic phases were dried (Na₂SO₄), and the conversion was measured by GC.

Representative Example for Amination Employing the LDH-System. Lyophilized cells of *E. coli* BL21 (DE3) containing overexpressed ω -transaminases (20 mg) were rehydrated in an Eppendorf tube (2 mL) in phosphate buffer (0.6 mL, 100 mM, pH 7, 1 mM PLP) for 30 min, at 30 °C, at 120 rpm on an orbital shaker turned 90° sideward. Phosphate buffer (0.4 mL, 100 mM, pH 7, 1 mM PLP), containing glucose (27.0 mg, final concentration 150 mM), alanine (22.3 mg, final concentration 250 mM), lactate dehydrogenase (90 U) and glucose dehydrogenase (30 U), NAD⁺ (final concentration 1 mM) as well as the ketone (final concentration 50 mM), was added. L- or D- alanine were used as amine donor employing either (S)- or (R)-selective ω TAs, respectively. DMSO was additionally added in selected experiments (15%, v v⁻¹). The mixture was shaken at 30 °C and 120 rpm. The reaction was quenched and worked up as described above.

Representative Example for Amination Employing the 2-Propylamine System. Lyophilized cells of *E. coli* BL21 (DE3)/pET-21a-ArRmut11- ω TA (20 mg) were rehydrated in a stock solution (800 μ L) containing PLP (0.5 mM) and 2propylamine (1 M), adjusted to pH 11 employing 3 N HCl, for 30 min, at 30 °C and 120 rpm in a glass vial (4 mL) on an orbital shaker. Then, DMSO (200 μ L) and ketone 1 (50 mM) were added. The vial was closed with a cap and shaken at 45 °C and 120 rpm in an upright position. The reaction was quenched and worked up as described above.

Determination of Optical Purity. The enantiomeric excess of amines 2 was analyzed by GC on a chiral phase (see Supporting Information) after derivatization to acetoamides, which was performed by adding 4-(N_1N -dimethylamino) pyridine (5 mg) dissolved in acetic anhydride (100 μ L). After washing with water and drying (Na_2SO_4) the *ee* value of the derivatized compound was measured. For more details see Supporting Information.

ASSOCIATED CONTENT

S Supporting Information

Information about the ω -transaminases employed, procedure for the amination of ketones **1a**–**f** employing (*S*)- and (*R*)selective ω -transaminases, chemo-enzymatic synthesis of 3amino-chromane from σ -hydroxy-benzaldehyde, and analytics. This material is available free of charge via the Internet at http://pubs.acs.org.

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Author Contributions

The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

Notes

The authors declare no competing financial interest.

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REFERENCES

(1) Turner, N. J.; Truppo, M. In Chiral Amine Synthesis: Methods, Developments and Applications; Nugent, T. C., Ed.; Wiley-VCH: Weinheim, Germany, 2010; pp 431–459.

(2) Nugent, T. C.; El-Shazly, M. Adv. Synth. Catal. 2010, 352, 753-819.

(3) Höhne, M.; Bornscheuer, U. T. ChemCatChem 2009, 1, 42-51.

(4) Aboul-Enein, H. Y.; Wainer, I. W. In *The Impact of Stereochemistry* on *Drug Development and Use*; Aboul-Enein, H. Y.; Wainer, I. W., Eds.; Wiley & Sons Inc.: New York, 1997.

(5) Ariëns, E. J.; Soudijin, W.; Timmermans, P. B. M. W. M. *Stereochemistry and Biological Activity of Drugs*; Blackwell Scientific Publishers: Oxford, U.K., 1983.

(6) Bertani, B.; di Fabio, R.; Micheli, F.; Tedesco, G.; Terreni, S. WO 2009/27294 A1 20090305, Glaxo Group Limited, 2009.

(7) Griebel, G.; Caroline, C.; Caroline, L.; Arvanitis, L. A. US2008/ 0033014 A1 20080207, Sanofi-Aventis, 2008.

(8) Van Ameringen, M.; Oakman, J.; Mancini, C.; Pipe, B.; Chung, H. Int. Clin. Psychopharmacol. **2004**, 24, 42–48.

(9) Smitt, J. A.; Ramaekers, J. G.; Kruizinga, M. J.; Van Boxtel, M. P.;
Vuurman, E. F.; Riedel, W. J. J. Psychopharmacol. 2002, 16, 207–214.
(10) Davidson, J. R.; Rothbaum, B. O.; van der Kolk, B. A.; Sikes, C.

R.; Farfel, G. M. Arch. Gen. Psychiatry 2001, 58, 485–492.

(11) Flament, M. F.; Lane, R. M.; Zhu, R.; Ying, Z. Int. Clin. Psychopharmacol. 1999, 14, 259–275.

(12) Han, Z.; Koenig, S. G.; Zhao, H.; Su, X.; Singh, S. P.; Bakale, R. P. Org. Process Res. Dev. 2007, 11, 726–730.

(13) Watts, R. L.; Jankovic, J.; Waters, C.; Rajput, A.; Boroojerdi, B.; Rao, J. *Neurology* **2007**, *68*, 272–276.

(14) Griebel, G.; Misslin, R.; Pawlowski, M.; Guardiola-Lemaître, B.; Guillaumet, G.; Bizot-Espiard, J. NeuroReport **1992**, *3*, 84–86.

(15) Beliaev, A.; Learmonth, D.; Smolenskaya, V. WO 2007/139413 A2 20071206, *Bial–Portela & C.A.*, 2007.

(16) Beliaev, A.; Learmonth, D. A.; Soares-da-Silva, P. J. Med. Chem. 2006, 3, 1191–1197.

(17) Arborelius, L.; Wallsten, C.; Ahlenius, S.; Svensson, T. H. Eur. J. Pharmacol. **1999**, 382, 133–138.

(18) Jerning, E.; Svantesson, G. T.; Mohell, N. Eur. J. Pharmacol. 1998, 360, 219–225.

(19) Sonesson, C.; Barf, T.; Nilsson, J.; Dijkstra, D.; Carlsson, A.; Svensson, K.; Smith, M. W.; Martin, I. J.; Duncan, J. N.; King, L. J.; Wikström, H. J. Med. Chem. **1995**, 38, 1319–1329.

(20) Zhuang, Z.-P.; Kung, M.-P.; Clarke, W.; Maayani, S.; Mu, M.; Kung, H. F. Chirality **1995**, *7*, 452–458.

(21) Karlsson, A.; Pettersson, C.; Sundell, S.; Arvidsson, L.-E.; Hacksell, U. Acta Chem. Scand. 1988, B 42, 231–236.

(22) Johansson, A. M.; Nilsson, J. L. G.; Karlén, A.; Hacksell, U.; Sanchez, D.; Svensson, K.; Hjorth, S.; Carlsson, A.; Sundell, S.; Kenne, L. J. Med. Chem. **1987**, 30, 1827–1837.

(23) Aaseng, J. E.; Melnes, S.; Reian, G.; Gautun, O. R. *Tetrahedron* **2010**, *66*, 9790–9797.

(24) Tschaen, D. M.; Abramson, L.; Cai, D.; Desmond, R.; Dolling, U.-H.; Frey, L.; Karady, S.; Shi, Y.-J.; Verhoeven, T. R. J. Org. Chem. **1995**, 60, 4324–4330.

(25) Buisson, D.; Cecchi, R.; Laffitte, J.-A.; Guzzi, U.; Azerad, R. Tetrahedron Lett. 1994, 35, 3091–3094.

(26) Kim, Y.; Park, J.; Kim, M.-J. ChemCatChem 2011, 3, 271-277.

(27) Kim, Y.; Park, J.; Kim, M.-J. Tetrahedron Lett. 2010, 51, 5581-5584.

(28) Thalén, L. K.; Zhao, D.; Sortais, J.-B.; Paetzold, J.; Hoben, C.; Bäckvall, J.-E. *Chem.—Eur. J.* **2009**, *15*, 3403–3410.

(29) Kim, M.-J.; Kim, W.-H.; Han, K.; Choi, Y. K.; Park, J. Org. Lett. 2007, 9, 1157–1159.

(30) Parvulescu, A. N.; Jacobs, P. A.; De Vos, D. E. Adv. Synth. Catal. 2008, 350, 113–121.

(31) Parvulescu, A. N.; Jacobs, P. A.; Vos, D. E. D. Chem.-Eur. J. 2007, 13, 2034-2043.

(32) Parvulescu, A. N.; Van der Eycken, E.; Jacobs, P. A.; De Vos, D. E. J. Catal. 2008, 255, 206–212.

(33) Wang, C.; Villa-Marcos, B.; Xiao, J. Chem. Commun. 2011, 47, 9773–9785.

(34) Zhou, X.-Y.; Bao, M.; Zhou, Y.-G. Adv. Synth. Catal. 2011, 353, 84-88.

(35) Guan, Z.-H.; Huang, K.; Yu, S.; Zhang, X. Org. Lett. 2009, 11, 481–483.

(36) Bernsmann, H.; van den Berg, M.; Hoen, R.; Minnaard, A. J.; Mehler, G.; Reetz, M. T.; De Vries, J. G.; Feringa, B. L. *J. Org. Chem.* **2005**, *70*, 943–951.

- (37) Tang, W.; Chi, Y.; Zhang, X. Org. Lett. 2002, 4, 1695–1698.
- (38) Patureau, F. W.; Worch, C.; Siegler, M. A.; Spek, A. L.; Bolm, C.; Reek, J. N. H. Adv. Synth. Catal. 2012, 354, 59-64.

(39) Jiang, X.-B.; Lefort, L.; Goudriaan, P. E.; de Vries, A. H. M.; van Leeuwen, P. W. N. M.; de Vries, J. G.; Reek, J. N. H. *Angew. Chem., Int. Ed.* **2006**, 45, 1223–1227.

- (40) Renaud, J. L.; Dupau, P.; Hay, A.-E.; Guingouain, M.; Dixneuf, P. H.; Bruneau, C. Adv. Synth. Catal. **2003**, 345, 230–238.
- (41) Mathew, S.; Yun, H. ACS Catal. 2012, 2, 993-1001.
- (42) Tufvesson, P.; Lima-Ramos, J.; Jensen, J. S.; Al-Haque, N.; Neto, W.; Woodley, J. M. *Biotechnol. Bioeng.* **2011**, *108*, 1479–1493.
- (43) Koszelewski, D.; Tauber, K.; Faber, K.; Kroutil, W. Trends Biotechnol. 2010, 28, 324-332.
- (44) Hailes, H. C.; Dalby, P. A.; Lye, G. J.; Baganz, F.; Micheletti, M.; Szita, N.; Ward, J. M. *Curr. Org. Chem.* **2010**, *14*, 1883–1893.
- (45) Ward, J.; Wohlgemuth, R. Curr. Org. Chem. 2010, 14, 1914–1927.
- (46) Humble, M. S.; Cassimjee, K. E.; Abedi, V.; Federsel, H.-J.; Berglund, P. ChemCatChem **2012**, *4*, 1167–1172.
- (47) Simon, R. C.; Grischek, B.; Zepeck, F.; Steinreiber, A.; Belaj, F.; Kroutil, W. Angew. Chem., Int. Ed. **2012**, *51*, 6713–6716.
- (48) Sattler, J. H.; Fuchs, M.; Tauber, K.; Mutti, F. G.; Faber, K.; Pfeffer, J.; Haas, T.; Kroutil, W. Angew. Chem., Int. Ed. 2012, 51, 9156–9159.
- (49) Truppo, M.; Janey, J. M.; Grau, B.; Morley, K.; Pollack, S.; Hughes, G.; Davies, I. *Catal. Sci. Technol.* **2012**, *2*, 1556–1559.
- (50) Shaheer Malik, M.; Park, E.-S.; Shin, J.-S. *Green Chem.* **2012**, *14*, 2137–2140.
- (51) Mutti, F. G.; Fuchs, C. S.; Pressnitz, D.; Turrini, N. G.; Sattler, J. H.; Lerchner, A.; Skerra, A.; Kroutil, W. *Eur. J. Org. Chem.* **2012**, 1003–1007.

(52) Cassimjee, K. E.; Svedendahl-Humble, M.; Land, H.; Abedi, V.; Berglund, P. Org. Biomol.Chem. 2012, 10, 5466-5470.

- (53) Seo, Y.-M.; Mathew, S.; Bea, H.-S.; Khang, Y.-H.; Lee, S.-H.; Kim, B.-G.; Yun, H. Org. Biomol. Chem. **2012**, 10, 2482–2485.
- (54) Humble, M. S.; Cassimjee, K. E.; Håkansson, M.; Kimbung, Y. R.; Walse, B.; Abedi, V.; Federsel, H.-J.; Berglund, P.; Logan, D. T. *FEBS J.* **2012**, *279*, 779–792.
- (55) Schätzle, S.; Steffen-Munsberg, F.; Thontowi, A.; Höhne, M.; Robins, K.; Bornscheuer, U. T. Adv. Synth. Catal. 2011, 353, 2439– 2445.
- (56) Mutti, F. G.; Fuchs, C. S.; Pressnitz, D.; Sattler, J. H.; Kroutil, W. Adv. Synth. Catal. 2011, 353, 3227–3233.
- (57) Bea, H.-S.; Park, H.-J.; Lee, S.-H.; Yun, H. Chem. Commun. 2011, 47, 5894–5896.
- (58) Matosevic, S.; Lee, G. J.; Baganz, F. J. Biotechnol. 2011, 155, 320–329.
- (59) Höhne, M.; Schätzle, S.; Jochens, H.; Robins, K.; Bornscheuer, U. T. Nat. Chem. Biol. **2010**, *6*, 807–813.
- (60) Smith, E. B.; Chen, B. H.; Hibbert, E. G.; Kaulmann, U.; Smithies, K.; Galman, J. L.; Baganz, F.; Dalby, P. A.; Hailes, H. C.; Lye,
- G. J.; Ward, J. M.; Woodley, J. M.; Micheletti, M. Org. Process Res. Dev. 2010, 14, 99–107.
- (61) Truppo, M. D.; Rozzell, J. D.; Moore, J. C.; Tuner, N. J. Org. Biomol. Chem. 2009, 7, 395–398.
- (62) Koszelewski, D.; Lavandera, I.; Clay, D.; Guebitz, G. M.;
 Rozzell, D.; Kroutil, W. Angew. Chem., Int. Ed. 2008, 47, 9337–9340.
 (63) Mutti, F. G.; Kroutil, W. Adv. Synth. Catal. 2012, 354, 3409–3413.
- (64) Savile, C. K.; Janey, J. M.; Mundorff, E. C.; Moore, J. C.; Tam, S.; Jarvis, W. R.; Colbeck, J. C.; Krebber, A.; Fleitz, F. J.; Brands, J.; Devine, P. N.; Huisman, G. W.; Hughes, G. J. *Science* **2010**, *329*, 305–309.
- (65) Shin, J.-S.; Yun, H.; Jang, J.-W.; Park, I.; Kim, B.-G. Appl. Microbiol. Biotechnol. 2003, 61, 463–471.

- (66) Kaulmann, U.; Smithies, K.; Smith, M. E. B.; Hailes, H. C.; Ward, J. M. Enzyme Microb. Technol. 2007, 41, 628-637.
- (67) Hanson, R. L.; Davis, B. L.; Chen, Y.; Goldberg, S. L.; Parker, W. L.; Tully, T. P.; Montana, M. A.; Patel, R. N. *Adv. Synth. Catal.* **2008**, 350, 1367–1375.
- (68) Park, E.; Kim, M.; Shin, J.-S. Adv. Synth.Catal. 2010, 352, 3391–3398.
- (69) Kawano, S.; Ito, N.; Yasohara, Y. WO 2007/139055 A1 20071206, Kaneka Corporation, Japan, 2007.
- (70) Pannuri, S.; Kamat, S. V.; Garcia, A. R. M. WO 2006/063336 A2 20060615, Cambrex North Brunswick Inc., 2006.
- (71) Learmonth, D. A.; Soares da Silva, P. M. V. A.; Beliaev, A. EP 1408038 A2 20040414, Portela & Ca. S. A., 2004.
- (72) Corey, E. J.; Wu, L. I. J. Am. Chem. Soc. 1993, 115, 9327–9328.
 (73) Alami, M.; Peyrat, J.-F.; Belachmi, L.; Brion, J.-D. Eur. J. Org. Chem. 2001, 4207–4212.
- (74) Koszelewski, D.; Göritzer, M.; Clay, D.; Seisser, B.; Kroutil, W. ChemCatChem 2010, 2, 73-77.